

# Lithium-Free Covalent Chemical Functionalization of Two-Dimensional Molybdenum Disulfide

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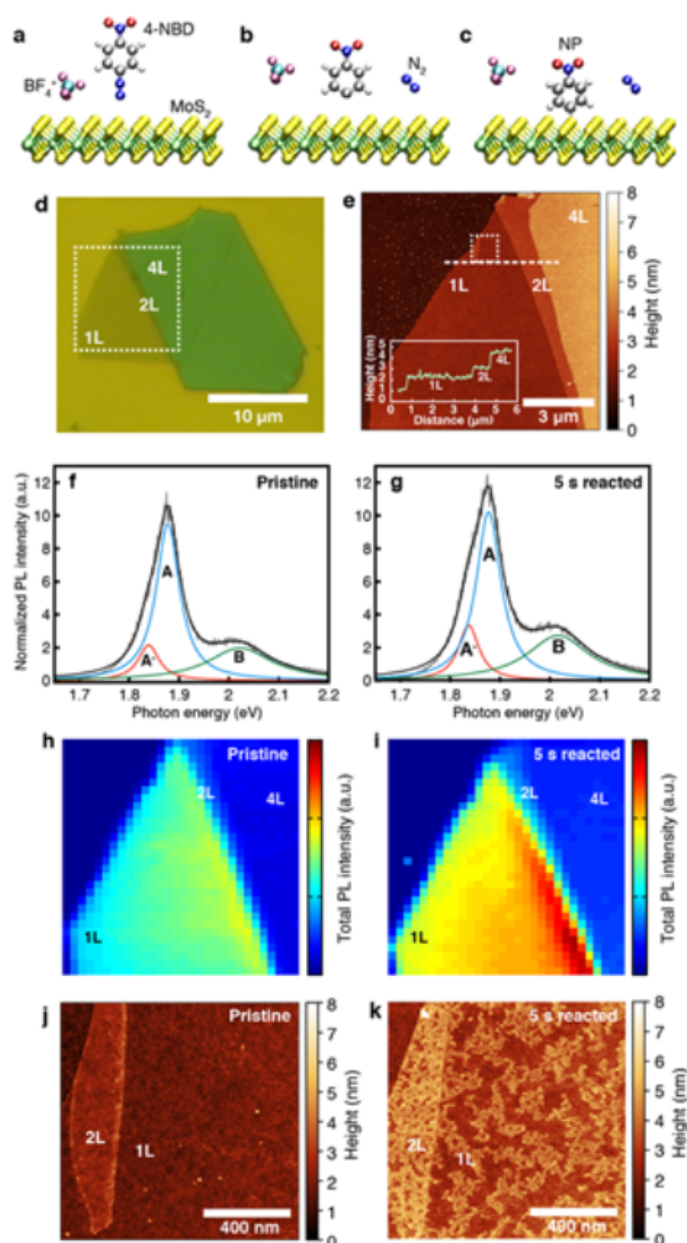
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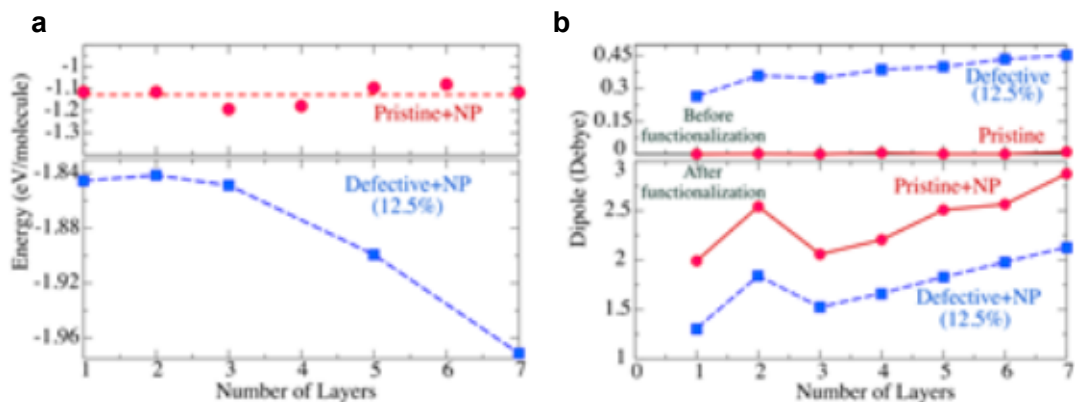
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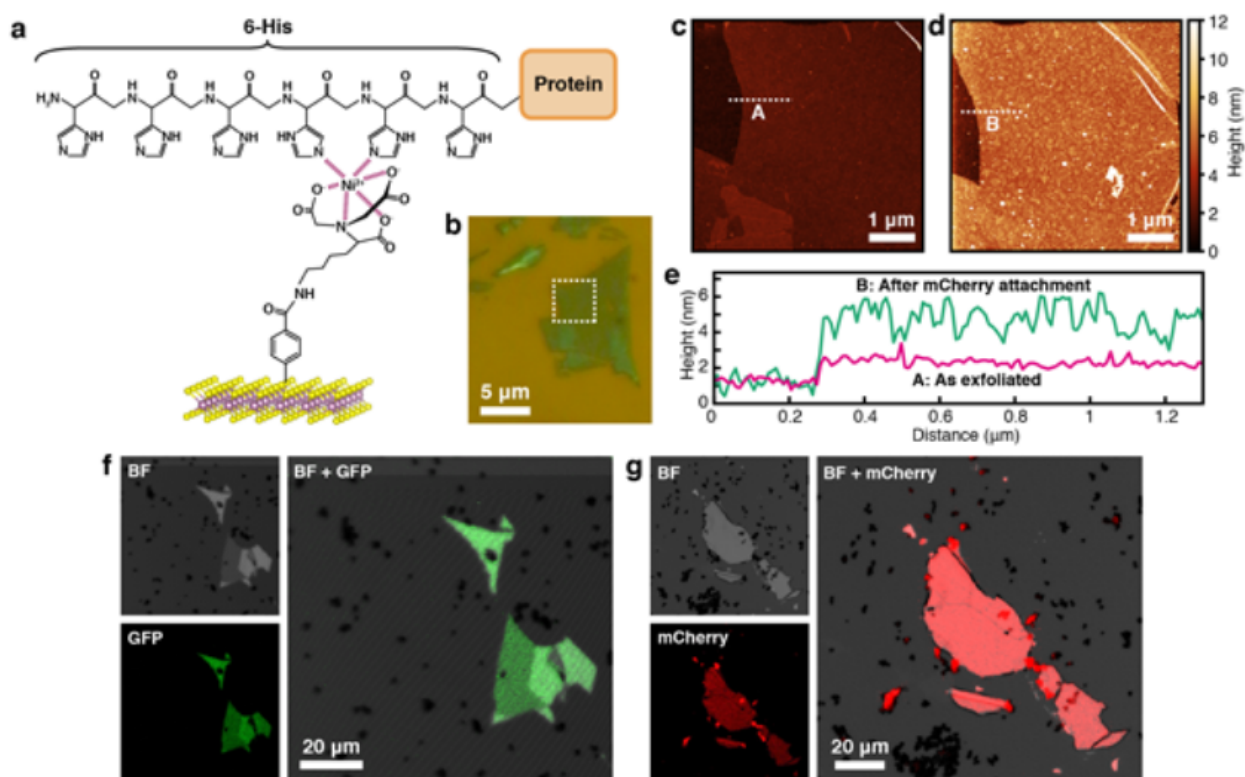
## Supplemental Figures for AVS Abstract



**Figure 1: Covalent functionalization of MoS<sub>2</sub>.** (a)-(c) Schematic illustrations of the functionalization reaction. The aryl diazonium salt 4-nitrobenzenediazonium (4-NBD) tetrafluoroborate (BF<sub>4</sub>) is dissolved in aqueous solution. Charges at the MoS<sub>2</sub> surface cause the diazonium group to break from the molecule as a nitrogen (N<sub>2</sub>) molecule. The resulting nitrobenzene radical forms a covalent bond to a sulfur atom on the surface, resulting in a nitrophenyl (NP) functional group. (d) Optical microscope image of mechanically exfoliated MoS<sub>2</sub> flake with monolayer (1L), bilayer (2L), and four-layer (4L) regions marked. (e) Atomic force microscope (AFM) image of the region marked by the dotted square in panel (d) of the pristine MoS<sub>2</sub> flake. Inset: Height profile along the dashed line. (f) Representative photoluminescence (PL) spectrum of 1L MoS<sub>2</sub> before reaction with 4-NBD, and (g) after 5 s reaction. Lorentzian lineshapes were used to fit the A and B exciton and A<sup>-</sup> trion peaks. The spectra are normalized to the height of the Raman peaks. (h) Spatial map of total integrated intensity of PL for MoS<sub>2</sub> in the region marked by the dotted square in panel (d) before reaction and (i) after 5 s reaction. The PL intensity is highest for monolayer MoS<sub>2</sub> due to its direct bandgap. (j) AFM image of the region in the dashed square in panel (e), with mainly 1L MoS<sub>2</sub> and a small region of 2L MoS<sub>2</sub>, before reaction and (k) after 5 s reaction. Many small protrusions in chain-like shapes are observed.



**Figure 2: Theoretical calculations of the effects of defects, coverage and number of layers on the functionalization of MoS<sub>2</sub>.** (a) Binding energy (eV per molecule) as a function of the number of layers for pristine (upper panel) and defective (lower panel) MoS<sub>2</sub> surface with 12.5% S-vacancy concentration. The dashed line in the pristine case is the average along the entire set of thicknesses, and in the defective case is the interpolation between the points. (b) Dipole moment at the MoS<sub>2</sub> surface as a function of the number of layers for pristine and defective (12.5% S-vacancies) MoS<sub>2</sub> before (upper panel) and after (lower panel) the functionalization.



**Figure 3: Attachment of active proteins to MoS<sub>2</sub>.** (a) Schematic of NTA-Ni-chelation attachment of poly-histidine (His) tagged protein, linked to MoS<sub>2</sub> surface via diazonium functionalization chemistry. The diazonium salt used here was 4-carboxybenzene diazonium (4-CBD) tetrafluoroborate. (b) Optical microscope image of mechanically exfoliated MoS<sub>2</sub> flakes featured in panels (c) and (d). (c) AFM image of pristine MoS<sub>2</sub> in the region indicated by the dashed square in panel (b). (d) AFM image in the same region as panel (c) after attachment of mCherry (red fluorescent protein) following initial 10 min functionalization with 4-CBD. (e) Height profiles along lines A and B in panels (c) and (d). (f-h) Bright field (BF), confocal fluorescence microscopy images in GFP (green) and mCherry (red) channels, and fluorescence images overlaid onto BF images, after protein attachment process. (f) GFP attachment. (g) mCherry attachment.