

Atomic Layer Etching

Room Samda Hall AB - Session ALE1-TuA

ALE Tools & ALE Modeling

Moderators: Satoshi Hamaguchi, Osaka University, Japan, Dmitry Suyatin, AlixLabs A.B.

1:30pm **ALE1-TuA-1 Isotropic and Anisotropic ALE: Tool Aspects, Processes, and Applications, Harm Knoops**, Oxford Instruments Plasma Technology, UK

INVITED
Applications' demands on control of etch depth, surface state, damage, and etching in 3D structures have caused atomic layer etching (ALE) to receive strong interest in recent years. One can distinguish ALE processes into two categories, isotropic and anisotropic processes. In this contribution, tool aspects, exemplary processes, and key applications for both isotropic and anisotropic ALE will be highlighted and, when appropriate, foundations and links between ALD and ALE will be addressed. Historically, tool designs for anisotropic ALE are built on those for reactive ion etching focusing on ion-energy control. Isotropic ALE processes are generally developed on ALD or ALD-like systems focusing on careful delivery and purging of precursor vapor at raised temperatures. To bring ALE to the next level, a closer look needs to be had at both the tool and process requirements. Among the general aspects important in the different steps in the ALE process are: control of plasma radical and ion fluxes and energies, delivery and removal of reactants and products from the chamber, process pressure and temperature (from ~350 °C down to cryogenic). For processes the discussion will range from BCl_3/Cl_2 -Ar anisotropic chemistries to SF_6 - AlMe_3 isotropic chemistries. Regarding applications, besides the well-known CMOS logic and 3D memory ones, power electronics, photonics, and quantum will be discussed. The aim being to highlight that control of interfaces and surfaces at the atomic scale is important even if the feature scale of the device is above ~100 nm. It is important to consider the different application spaces for ALE as this will also shape the requirements on tools and processes.

2:00pm **ALE1-TuA-3 Study on Plasma Induced Damaged Layer Formation Using Molecular Dynamics, Junghwan Um, Sung-Il Cho**, Samsung Electronics Co., Republic of Korea

Plasma etching is a key technology in semiconductor fabrication, enabling precise patterning at the nanoscale. However, plasma-surface interactions often lead to the formation of a damage layer, which can negatively impact material properties and device performance. Understanding the mechanisms of damage layer formation at the atomic scale is crucial for optimizing plasma etching conditions and minimizing unwanted effects.

In this study, we employ molecular dynamics (MD) simulations to investigate the fundamental processes involved in plasma-induced damage layer formation. The simulations focus on the effects of ion energy, ion species and surface chemistry on the evolution of the damage layer. By simulating energetic ion bombardment and its interactions with the substrate, we analyze structural modifications. Our results show that higher ion energy leads to deeper penetration of ions into the substrate, increasing the damage layer thickness. Low-energy ions, on the other hand, primarily cause surface modifications without significant subsurface damage. The choice of ion species also plays a critical role, with heavier ions inducing more structural disorder compared to lighter ions due to their higher momentum transfer. Furthermore, we examine the role of reactive species in plasma etching, such as fluorine (F) or chlorine (Cl), which contribute to both material removal and chemical modification of the substrate. The competition between physical sputtering and chemical etching is analyzed to determine the optimal conditions for achieving a damage-free etching process. By comparing MD simulation results with experimental findings, we establish a comprehensive understanding of plasma-induced damage mechanisms. Our study provides valuable insights into designing advanced plasma etching processes with minimized damage, which is essential for next-generation semiconductor devices. These findings can guide the development of new plasma processing strategies, including low-damage etching techniques and optimized plasma parameters for emerging nanofabrication applications.

2:15pm **ALE1-TuA-4 Theoretical Analysis on Crystalline Phase-Dependent Surface Fluorination of HfO_2 for Atomic Layer Etching, Sujin Kwon, Bonggeun Shong**, Hongik University, Republic of Korea

Hafnium oxide (HfO_2) thin films have garnered significant attention in microelectronics industry due to its high dielectric constant, and more
Tuesday Afternoon, June 24, 2025

recently, the ferroelectric (FE) properties of hafnium zirconium oxide (HZO). For applications of HfO_2 thin films, precise control of the thickness is necessary, and thus atomic layer deposition (ALD) and atomic layer etching (ALE) processes of HfO_2 can be crucial. Several ALE chemistry of HfO_2 are known, and most of them include fluorination of the surface using HF. While ALE is often believed to reduce roughness, some previous experimental reports indicate that the surface roughness of HZO thin films can increase after ALE [1]; additionally, the etch rates in ALE of HfO_2 and ZrO_2 are dependent on their crystallinity [2], which can be related to the propensity toward surface fluorination [3]. In this study, we investigate the variation of surface fluorination of HfO_2 according to its crystalline phases. The HF-based surface fluorination of four representative phases of HfO_2 – monoclinic, tetragonal, orthorhombic, and cubic – is explored. Using machine-learning interatomic potential (MLIP) calculations, Natarajan-Elliott (N-E) analyses on spontaneous etch (SE) versus self-limiting (SL) conditions of surface fluorination are performed [4]. Based on current results, it can be inferred that the variation in fluorination behaviors across different crystallinity and phases of the polycrystalline HfO_2 thin films may lead to increased surface roughness after ALE due to differences in etch rates.

Acknowledgments. This work was supported by Samsung Electronics.

References [1] Appl. Phys. Lett. 120, 122901 (2022), [2] J. Vac. Sci. Technol. A 38, 022608 (2020), [3] J. Vac. Sci. Technol. A 40, 022604 (2022), [4] Chem. Mater. 32(8), 3414-3426 (2020)

2:30pm **ALE1-TuA-5 Removal Reaction Mechanisms During Thermal Atomic Layer Etching of Aluminum Oxide: A First-Principles Study, Khabib Khumaini, Gyejun Cho, Hye-Lee Kim, Won-Jun Lee**, Sejong University, Republic of Korea

Understanding the mechanism of atomic layer etching (ALE) is essential for process design and optimization. Thermal ALE of metal oxides typically involves surface fluorination followed by ligand exchange reactions to remove the fluorinated layer. In the removal step, surface fluorine groups are usually replaced by methyl groups or chlorine atoms to increase the volatility of the modified substrate using metal precursors. Surface reactions during the removal step of ALE aluminum oxide (Al_2O_3) using various metal precursors have been reported using *in situ* characterizations [1]. However, the investigation of the atomic-scale mechanism remains limited. Therefore, we performed density functional theory (DFT) calculations to study the removal reaction during ALE of Al_2O_3 . Since the fluorination step produces an amorphous aluminum fluoride (a- AlF_3) layer, we constructed an amorphous substrate model by the melt-quench method and optimized the surface group density. Removal reactions with $\text{Al}(\text{CH}_3)_3$, AlCl_3 , $\text{AlCl}(\text{CH}_3)_2$, SiCl_4 , and TiCl_4 were then simulated. At 250°C, the removal of a- AlF_3 by $\text{Al}(\text{CH}_3)_3$ and AlCl_3 , releasing $\text{Al}_2\text{F}_2(\text{CH}_3)_4$ and $\text{Al}_2\text{F}_2\text{Cl}_4$, occurs spontaneously with low activation energies of 1.08 and 0.85 eV, respectively. $\text{AlCl}(\text{CH}_3)_2$ preferentially removes a- AlF_3 with activation energies of 0.54–0.74 eV, which is lower than $\text{Al}(\text{CH}_3)_3$ and AlCl_3 . Conversely, reactions with SiCl_4 or TiCl_4 release only SiFCl_3 or TiFCl_3 molecules with no release of aluminum-containing species, indicating that a- AlF_3 etching does not occur at 250°C. These results are in agreement with the experimental observations. Our DFT calculation results indicate that the substitution of a methyl group with a chlorine atom in the precursor increases the reactivity, while the substitution of aluminum with silicon and titanium significantly decreases both reactivity and spontaneity. These results will contribute to the rational design of precursors for thermal ALE processes.

[1] J.W. Clancey et al., J. Phys. Chem. C **124** (2020) 287–299.

2:45pm **ALE1-TuA-6 Multiscale Modeling of Gallium Nitride Atomic Layer Etching in Chlorinated Plasmas: A Combined Dynamic Global Model, Ab-initio and Kinetic Monte Carlo Approaches, Tojo Rasanarivo, Cédric Mannequin, Isabelle Braems**, Institut des Matériaux de Nantes Jean Rouxel, France; Fabrice Roqueta, Mohamed Boufnichel, STMicroelectronics, France; Ahmed Rhallabi, Institut des Matériaux de Nantes Jean Rouxel, France

Plasma Atomic Layer Etching (ALE) is a cyclic etching process for which one cycle relies on two self-limited half-reactions, separated by purges. An ideal ALE cycle comprises an adsorption step to modify the outermost surface layer, followed by an activation step to selectively remove the aforementioned modified layer without etching the underlying non-modified layers, thereby achieving atomic-scale resolution. The ALE of GaN using alternating Cl_2 and Ar plasma for the adsorption and activation steps, respectively, has been the subject of extensive research [1]. These studies suggest that the adsorption step relies on surface modification by chlorine

radicals, while the activation step is achieved by selectively controlling the energy of ionic bombardment. However, the majority of ALE research has been conducted through experimental approaches, with some modeling studies employing molecular dynamics for common materials such as silica-based substrates [2].

In this study, we propose a multiscale kinetic model of GaN ALE in chlorinated plasmas. The temporal evolution of precursors fluxes during ALE cycles is provided by a dynamic global model of chlorinated plasmas [3]. The interactions between chlorine species and the GaN surface are investigated through the use of Density Functional Theory (DFT), while ion bombardments are consequently calculated using the Stopping and Range of Ions in Matter software (SRIM) considering the ion energy and angular distributions induced by the sheath at the interface between the plasma and the surface. Thereafter, those results are implemented in our new kinetic Monte-Carlo etching model with an atomic scale description. The simulation results demonstrate the effects of machine parameters (pulse lengths, flow rates, pressure, RF power) on the plasma composition, the etched depth and the etched GaN surface morphology at each ALE cycle.

Our approach is less time-consuming than molecular dynamics or DFT-only methods for atomic layer etching processes while following entire processes. Its suitability to a wide range of ALE recipes, for instance purge-free processes, could offer insights for process optimization.

References

- [1] Mannequin, Vallée, Akimoto, Chevolleau, et al., JVST A, (2020)
- [2] Kounis-Melas, Vella, Panagiotopoulos, Graves, JVST A, (2025)
- [3] Rasoanarivo, Mannequin, Roqueta, Boufnichel, Rhallabi, JVST A, (2024)

3:00pm ALE1-TuA-7 Utilization of Molecular Dynamics Simulations and a Reduced Order Model to Analyze the Atomic Layer Etching Window of the Si-Cl₂-Ar⁺ System, Joseph Vella, TEL Technology Center, America, LLC, USA; David Graves, Department of Chemical and Biological Engineering, Princeton University

Atomic-layer etching (ALE) processes are often characterized by the ALE window. The ALE window is a range of ion energies where the amount of substrate stays constant as a function of the ion energy. The Si-Cl₂-Ar⁺ system is often used as an example to illustrate concepts of ALE, including the ALE window.[1] Despite this, when examining the literature, properties of the ALE window for this system remain obscure. For example, Kim et al.[2] studied Si-Cl₂-Ar⁺ ALE and report that the ALE window should be below 40 eV. On the other hand, Park et al.[3] report the ALE window as being from 70 to 90 eV. Still others report an Ar⁺ ion energy of 50 eV as being within the ALE window.[4] In this work, we aim to resolve these contradictory reports by studying the Si-Cl₂-Ar⁺ ALE with classical molecular dynamics (MD) simulations and a reduced order model (ROM), also known as a transient site balance model.[5] The MD results show that the range of Ar⁺ ion energies where the amount of Si etched per cycle (EPC) does not change is from 15eV to 20 eV, which is very narrow. The EPC in this region is also less than one atomic layer, because atomic Cl sputtering is significant. Despite this, it is believed this is where the ALE window is located. The MD simulations also show that a large ion fluence (roughly $4.2 \cdot 10^{16}$ ions/cm² for 15 eV ions) is required to remove all Cl from the near surface region. Using the ROM, parameters can be varied to observe their effect on properties of the ALE window. For example, by increasing the threshold sputtering energy of Si, the width of ALE window can be increased. While this study focuses on the relatively simple Si-Cl₂-Ar⁺ system, it is clear learnings from this study can be extended to general ALE processes.

References

- [1] T. Lill, "Atomic Layer Processing: Semiconductor Dry Etching Technology" (Wiley-VCH, Weinheim, 2021).
- [2] B. Kim, S. Chung, and S. M. Cho, "Layer-by-layer Etching of Cl-adsorbed Silicon Surfaces by Low Energy Ar⁺ Ion Irradiation", Appl. Surf. Sci. 2002, 187, 124-129.
- [3] S. Park, K. Min, B. Yoon, D. Lee, and G. Yeom, "Precise Depth Control of Silicon Etching using Chlorine Atomic Layer Etching" Jpn. J. Appl. Phys. 2005, 44, 389-393.
- [4] K. J. Kanarik, T. Lill, E. A. Hudson, S. Sriraman, S. Tan, J. Marks, V. Vahedi, and R. A. Gottscho, "Overview of Atomic Layer Etching in the Semiconductor Industry", J. Vac. Sci. Technol. A, 2015, 33, 020802.
- [5] J. R. Vella, Q. Hao, M. A. I. Elgarhy, V. M. Donnelly, and D. B. Graves, "A Transient Site Balance Model for Atomic Layer Etching", Plasma Sources Sci. Technol., 2024, 33, 075009.

3:15pm ALE1-TuA-8 Characteristics of the Power Delivery System of the Transformer-Coupled Plasma Source for Remote Plasma Process in Semiconductor Manufacturing, Tae S. Cho, Hakmin Kim, Giwon Shin, Jaehoon Choi, Sooyoung Hwang, Jihyun Kim, Wonik IPS, Republic of Korea

As the semiconductor industry progresses with sub-10 nm features, 3D stacked architectures, and intricate gate-all-around structures, the demand for remote plasma technologies has grown significantly. The requirements for the remote plasma are to generate a high density of radicals at low power and deliver them to the wafer processing region with minimal loss during transport, while preventing damage to the wafer from ion bombardment. "We have conducted an analysis of the power delivery system of the transformer-coupled plasma (TCP), which is one of the most commonly used plasma sources that meets these requirements. Since the power efficiency of the TCP directly influences the plasma properties, understanding the power transfer characteristics is crucial for improving process stability and uniformity."

The model was expressed as a function of the number of ferrite cores and dielectric breaks, that are the primary components of the TCP. The model was experimentally validated by varying the number of ferrites and breaks in the TCP, and it was confirmed that the experimental results were in good agreement with the values predicted by the model. Therefore, the proposed model was able to effectively predict the characteristics of the power delivery system of the TCP source. In future work, we aim to derive a more accurate equivalent model and a generalized equation by incorporating variables related to the plasma properties

Author Index

Bold page numbers indicate presenter

— B —

Boufnichel, Mohamed: ALE1-TuA-6, 1
Braems, Isabelle: ALE1-TuA-6, 1

— C —

Cho, Gyejun: ALE1-TuA-5, 1
Cho, Sung-Il: ALE1-TuA-3, 1
Cho, Tae S.: ALE1-TuA-8, 2
Choi, Jaehoon: ALE1-TuA-8, 2

— G —

Graves, David: ALE1-TuA-7, 2

— H —

Hwang, Sooyoung: ALE1-TuA-8, 2

— K —

Khumaini, Khabib: ALE1-TuA-5, 1
Kim, Hakmin: ALE1-TuA-8, 2
Kim, Hye-Lee: ALE1-TuA-5, 1
Kim, Jihyun: ALE1-TuA-8, 2
Knoops, Harm: ALE1-TuA-1, 1
Kwon, Sujin: ALE1-TuA-4, 1

— L —

Lee, Won-Jun: ALE1-TuA-5, 1

— M —

Mannequin, Cédric: ALE1-TuA-6, 1

— R —

Rasoanarivo, Tojo: ALE1-TuA-6, 1
Rhallabi, Ahmed: ALE1-TuA-6, 1
Roqueta, Fabrice: ALE1-TuA-6, 1

— S —

Shin, Giwon: ALE1-TuA-8, 2
Shong, Bonggeun: ALE1-TuA-4, 1

— U —

Um, Junghwan: ALE1-TuA-3, 1

— V —

Vella, Joseph: ALE1-TuA-7, 2