



# Gas-Phase Etching Mechanism of Amorphous Hydrogenated Silicon Nitride by Hydrogen Fluoride: A Theoretical Study

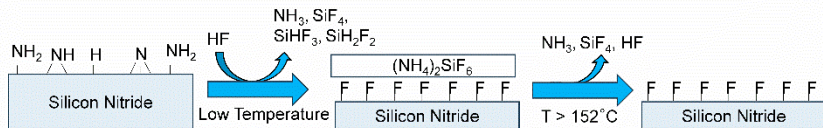
Khabib Khumaini,<sup>1,2</sup> Yewon Kim,<sup>1</sup> Romel Hidayat,<sup>1</sup> Tanzia Chowdhury,<sup>1</sup> Hye-Lee Kim,<sup>1</sup> Byungchul Cho,<sup>3</sup> Sangjoon Park,<sup>3</sup> and Won-Jun Lee,<sup>1\*</sup>

<sup>1</sup>Department of Nanotechnology and Advanced Materials Engineering, Sejong University, Seoul Republic of Korea <sup>2</sup>Department of Chemistry, Universitas Pertamina, Jakarta, Indonesia

<sup>3</sup>Wonik IPS, Pyeongtaek, Republic of Korea

\*Corresponding author: [wjlee@sejong.ac.kr](mailto:wjlee@sejong.ac.kr)

## Summary – Reaction Mechanism



## Simulation of Fluorination

### a. First Fluorination

Pathway	Reactions	$E_{\text{Pby}}$	$\Delta E$	$E_a$
P1a	SiNH <sub>2</sub> * + HF → SiF* + NH <sub>3</sub>	-1.09	-0.24	0.45
P1b	SiH* + HF → SiF* + H <sub>2</sub>	-0.83	-0.63	2.10
P1c	Si-NH*-Si + HF → SiF* + SiNH <sub>2</sub> *	-0.91	-1.40	0.02
P1d	Si-N-Si + HF → SiF* + SiNH*	-0.90	-0.97	0.72

### b. Second Fluorination

Pathway	Reactions	$E_{\text{Pby}}$	$\Delta E$	$E_a$
P2a	Si(NH <sub>2</sub> )F* + HF → SiF <sub>2</sub> * + NH <sub>3</sub>	-0.54	-0.76	0.68
P2b	SiHF* + HF → SiF <sub>2</sub> * + H <sub>2</sub>	-0.42	-0.65	1.80
P2c	SiF*-NH*-Si + HF → SiF <sub>2</sub> * + SiNH <sub>2</sub> *	-0.35	-0.24	0.79
P2d	SiF*-N-Si + HF → SiF <sub>2</sub> * + SiNH*	-0.51	-0.96	0.75
P2e	Si-N-SiH <sub>2</sub> F* + HF → SiH <sub>2</sub> F <sub>2</sub> + SiNH*	-0.51	-1.29	0.81
P2f	SiF*-Si + HF → SiF <sub>2</sub> * + SiH*	-0.25	-1.31	0.79

### c. Third Fluorination

Pathway	Reactions	$E_{\text{Pby}}$	$\Delta E$	$E_a$
P3a	Si(NH <sub>2</sub> )F <sub>2</sub> * + HF → SiF <sub>3</sub> * + NH <sub>3</sub>	-0.64	-0.42	0.88
P3b	SiHF <sub>2</sub> * + HF → SiF <sub>3</sub> * + H <sub>2</sub>	-0.29	-1.08	1.54
P3c	SiF <sub>2</sub> *-NH*-Si + HF → SiF <sub>3</sub> * + SiNH <sub>2</sub> *	-0.37	-0.67	0.74
P3d	SiF <sub>2</sub> *-N-Si + HF → SiF <sub>3</sub> * + SiNH*	-0.30	-0.56	0.90
P3e	Si-N-SiHF <sub>2</sub> * + HF → SiHF <sub>3</sub> + SiNH*	-0.39	-1.13	0.96

### d. Fourth Fluorination

Pathway	Reactions	$E_{\text{Pby}}$	$\Delta E$	$E_a$
P4	Si-NH*-SiF <sub>3</sub> * + HF → SiF <sub>4</sub> + SiNH <sub>2</sub> *	-0.32	-0.77	0.53

## Simulation of Salt Formation

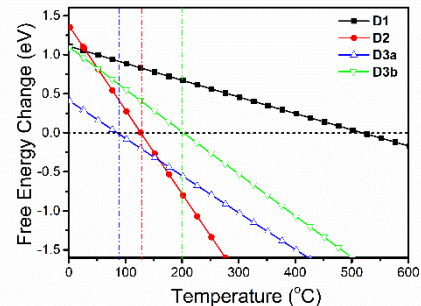
### a. Salt Formation

Reaction	Reactions	$\Delta E$	$E_a$
1	SiF <sub>4</sub> + NH <sub>3</sub> F → NH <sub>4</sub> <sup>+</sup> + SiF <sub>5</sub> <sup>-</sup>	-0.06	0.40
2	NH <sub>4</sub> <sup>+</sup> + SiF <sub>5</sub> <sup>-</sup> + NH <sub>4</sub> F → 2 NH <sub>4</sub> <sup>+</sup> + SiF <sub>6</sub> <sup>2-</sup>	-0.31	0.07

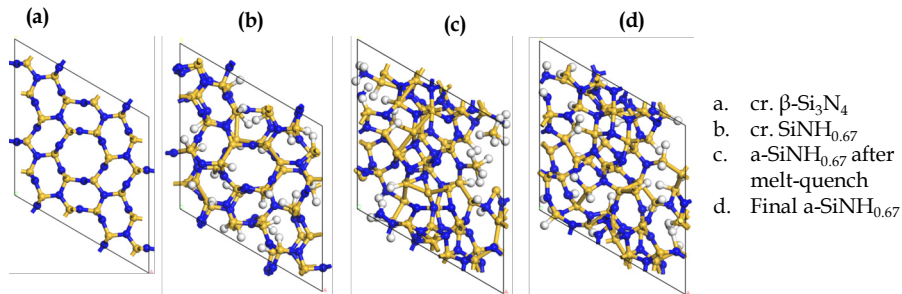
### b. Salt Desorption

Pathway	Reaction	$E_{\text{des}}$	$T_s$
D1	(NH <sub>4</sub> ) <sub>2</sub> SiF <sub>6</sub> (surf) → (NH <sub>4</sub> ) <sub>2</sub> SiF <sub>6</sub> (g)	1.98	526
D2	(NH <sub>4</sub> ) <sub>2</sub> SiF <sub>6</sub> (surf) → 2 NH <sub>3</sub> + 2 HF (g) + SiF <sub>4</sub> (g)	4.79	152
D3a	(NH <sub>4</sub> ) <sub>2</sub> SiF <sub>6</sub> (surf) → NH <sub>3</sub> F <sub>2</sub> (surf) + NH <sub>3</sub> (g) + SiF <sub>4</sub> (g)	1.97	102
D3b	NH <sub>5</sub> F <sub>2</sub> (surf) → NH <sub>3</sub> (g) + 2HF (g)	2.81	200

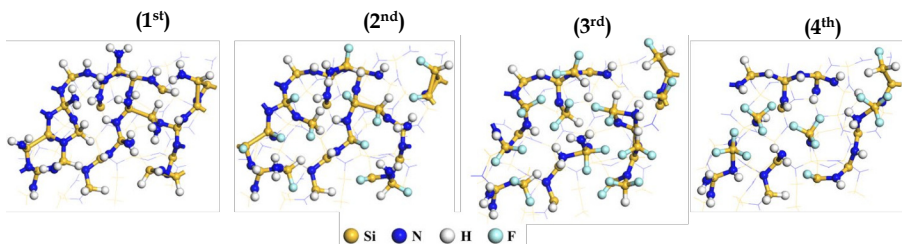
## Simulation of Salt Desorption Temperature



## Construction of the amorphous model



## Construction of surface models with different degrees of fluorination



Legend: Si (yellow), N (blue), H (white), F (light blue)