Thermal Atomic Layer Etching of Zinc Sulfide (ZnS) Using Sequential AI(CH₃)₃ and HF Exposures

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Figure 1. Estimated mechanism of ALE ZnS. (a) Conversion of ZnS by the reaction with $AI(CH_3)_3$, (b) fluorination of AI_2S_3 by the reaction with HF, and (c) ligand exchange of AIF_3 and conversion of ZnS by the reaction with $AI(CH_3)_3$.



Figure 2. Time-resolved quadruple mass spectrometer result of ALE ZnS during Al(CH₃)₃ and HF exposure at 275 °C